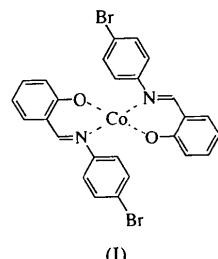


- Muir, J. A., Cuadrado, S. I. & Muir, M. M. (1991). *Acta Cryst. C* **47**, 1072–1074.
 Parish, R. V. & Cottriall, S. M. (1987). *Gold Bull.* **20**, 3–12.
 Rauhut, M. M., Hechenbleikner, I., Currier, H. A., Schaefer, F. C. & Wystrach, V. P. (1959). *J. Am. Chem. Soc.* **81**, 1103–1106.
 Sheldrick, G. M. (1990). *Acta Cryst. A* **46**, 467–473.
 Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
 Stinson, S. C. (1989). *Chem. Eng. News.* **67**, 37–38.
 Uson, R., Laguna, A. & Laguna, M. (1989). *Inorg. Synth.* **26**, 85–87.
 Walz, D. T., DiMartino, M. J. & Sutton, B. M. (1974). *Anti-inflammatory Agents*, p. 217. New York: Academic Press.

the structure of the title complex, (I), formed by the reaction of *N*-(4-bromophenyl)salicylaldimine with cobalt(II) acetate, was determined.



Acta Cryst. (1996). **C52**, 553–554

trans-Bis[*N*-(4-bromophenyl)salicylaldoiminato]cobalt(II)

AYHAN ELMALI,^a YALCIN ELERMAN,^a INGRID SVOBODA^{b*} AND HARTMUT FUESS^b

^aDepartment of Engineering Physics, Faculty of Sciences, University of Ankara, 06100 Besevler, Ankara, Turkey, and

^bStrukturforschung, FB Materialwissenschaft, Technische Hochschule Darmstadt, Petersenstr. 20, 64287 Darmstadt, Germany. E-mail: deos@hrzpub.th-darmstadt.de

(Received 11 October 1994; accepted 25 August 1995)

Abstract

In the title compound, *trans*-bis[2-(4-bromophenyliminomethyl)phenolato-*N*,*O*]cobalt(II), [Co(C₁₃H₉BrNO)₂], the Co atom is situated on a twofold axis and the geometry around the metal centre can be described as distorted tetrahedral with a *trans* conformation. The Co—O and Co—N bond lengths are 1.891 (3) and 2.006 (4) Å, respectively.

Comment

Because of their preparative accessibility and structural variability a great number of Schiff base complexes of cobalt have been subject to extensive studies. Tetra-coordinate Schiff base metal complexes may form *trans* and *cis* planar or tetrahedral structures. A strictly planar or slightly distorted *trans* configuration is characteristic for transition metal complexes of cobalt(II) with a CoN₂O₂ coordination sphere (Garnovskii, Nivorozhkin & Minkin, 1993). Structural and molecular properties of related Co^{II} complexes have been reported (Hennig, 1971; Panova, Vikulova & Popotov, 1980; Kogan, Kharabov, Osipov & Kochin, 1981).

In the course of a systematic investigation of the coordination spheres of Co^{II} in Schiff base complexes

The Co atom lies on a twofold axis and is coordinated by two O and two N atoms. The distortion from idealized geometry is due to the O1—Co—N1 angle of 96.75 (4)°, which is smaller than the O1—Co—N1 angle of 111.59 (14)°. The Co—O and Co—N distances found in the structure agree with the values in other tetrahedral complexes.

The phenyl rings show small distortions from ideal geometry, with the C—C distances for the phenyl rings close to the expected value for aromatic rings [1.394 (5) Å; Sutton, 1965]. The bond lengths between the N1 and C1 atoms are typical of C=N double-bond lengths [1.287 (6) Å]. The bond lengths Br1—C11 of 1.892 (5), O1—C3 of 1.319 (5) and N1—C8 of 1.428 (6) Å are within expected ranges.

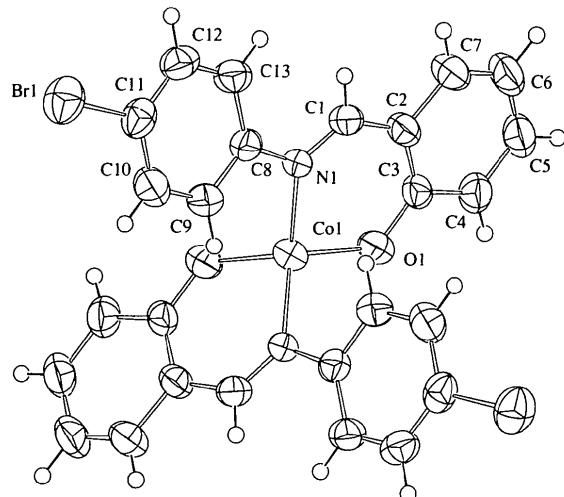
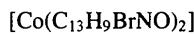


Fig. 1. ORTEP view (Johnson, 1965) of the molecular structure of the title compound. Displacement ellipsoids are plotted at the 50% probability level.

Experimental

0.5 mmol of [Co(AcO)₂]₂·4H₂O in methanol was added to 1.0 mmol of *N*-(4-bromophenyl)salicylaldimine in acetonitrile with stirring and heating. The mixture was refluxed for 6 h. The red crystals that formed were washed twice with ethanol.

Crystal data*M_r* = 609.18

Monoclinic

*C2/c**a* = 21.839 (9) Å*b* = 8.995 (5) Å*c* = 11.823 (5) Å β = 95.91 (4)°*V* = 2310.2 (2) Å³*Z* = 4*D_x* = 1.751 Mg m⁻³

Mo *K* α radiation
 λ = 0.71069 Å
 Cell parameters from 24 reflections
 θ = 7.26–19.05°
 μ = 4.229 mm⁻¹
 T = 298 (2) K
 Prism
 0.20 × 0.15 × 0.05 mm
 Red

O1—C3	1.319 (5)	C9—C10	1.379 (7)
C1—C2	1.431 (7)	C10—C11	1.380 (7)
C2—C7	1.414 (6)	C11—C12	1.377 (7)
C2—C3	1.427 (7)	C12—C13	1.374 (7)
C3—C4	1.396 (6)		
O1'—Co—O1	121.1 (2)	N1—C1—C2	128.6 (4)
O1'—Co—N1'	111.59 (14)	O1—C3—C4	118.6 (4)
O1—Co—N1	96.75 (14)	O1—C3—C2	122.9 (4)
N1'—Co—N1	120.8 (2)	C13—C8—N1	124.8 (4)
C1—N1—C8	120.4 (4)	C9—C8—N1	116.9 (4)
C1—N1—Co	119.5 (3)	C12—C11—Br1	120.5 (4)
C8—N1—Co	119.9 (3)	C10—C11—Br1	119.0 (4)
C3—O1—Co	126.3 (3)		

Symmetry codes: (i) $-x, y, \frac{1}{2} - z$.

All the phenyl rings were refined without constraints. Values of the distances and angles in the rings show no significant differences from those of an ideal benzene ring. H atoms were positioned geometrically.

Data collection: *DIF4* (Stoe & Cie, 1991a). Cell refinement: *DIF4*. Data reduction: *REDU4* (Stoe & Cie, 1991b). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *Xtal3.2* (Hall, Flack & Stewart, 1992). Software used to prepare material for publication: *SHELXL93*.

AE is grateful to the Scientific and Technical Research Council of Turkey (TUBITAK) and the Deutsche Forschungsgemeinschaft (DFG) for support.

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates and complete geometry have been deposited with the IUCr (Reference: AB1229). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Garnovskii, A. D., Nivorozhkin, A. L. & Minkin, V. I. (1993). *Coord. Chem. Rev.* **126**, 1–69.
 Hall, S. R., Flack, H. D. & Stewart, J. M. (1992). Editors. *Xtal3.2 Reference Manual*. Universities of Western Australia, Australia, Geneva, Switzerland, and Maryland, USA.
 Hennig, H. (1971). *Z. Chem.* **11**, 81–86.
 Johnson, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
 Kogan, V. A., Kharabov, N. N., Osipov, O. A. & Kochin, G. G. (1981). *Zh. Strukl. Khim.* **22**, 126–132.
 North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst. A24*, 351–359.
 Panova, G. V., Vikulova, N. K. & Popotov, V. M. (1980). *Usp. Khim.* **49**, 1234–1240.
 Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. University of Göttingen, Germany.
 Sheldrick, G. M. (1993). *SHELXL93. Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
 Stoe & Cie (1991a). *DIF4. Stoe Four-Circle Diffractometer Control Program*. Version 7.08. Stoe & Cie, Darmstadt, Germany.
 Stoe & Cie (1991b). *REDU4. Stoe Four-Circle Data Reduction Program*. Version 7.08. Stoe & Cie, Darmstadt, Germany.
 Sutton, L. E. (1965). *Chem. Soc. Spec. Publ. Suppl.* **18**, 516s.

Data collection

Stoe Stadi4 four-circle diffractometer

 $\omega/2\theta$ scansAbsorption correction:
 ψ scan (North, Phillips & Mathews, 1968) T_{\min} = 0.59, T_{\max} = 0.99

3385 measured reflections

1604 independent reflections

*Refinement*Refinement on F^2 $R(F)$ = 0.0334 $wR(F^2)$ = 0.0924 S = 1.057

1603 reflections

150 parameters

 $w = 1/[\sigma^2(F_o^2) + (0.0420P)^2 + 3.6412P]$ where $P = (F_o^2 + 2F_c^2)/3$

1210 observed reflections

[$I > 2\sigma(I)$] R_{int} = 0.0341 θ_{\max} = 23.02° h = −24 → 24 k = 0 → 9 l = −13 → 13

3 standard reflections

frequency: 120 min

intensity decay: 0.9%

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)

$$U_{\text{eq}} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* a_i a_j$$

	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Co	0	0.51515 (10)	1/4	0.0484 (3)
Br1	−0.15744 (3)	1.07167 (7)	−0.10946 (5)	0.0717 (2)
N1	0.0262 (2)	0.6253 (4)	0.1155 (3)	0.0406 (9)
O1	0.07476 (14)	0.4118 (4)	0.2873 (3)	0.0542 (9)
C1	0.0776 (2)	0.5921 (5)	0.0775 (4)	0.048 (1)
C2	0.1242 (2)	0.4912 (5)	0.1241 (4)	0.047 (1)
C3	0.1219 (2)	0.4070 (5)	0.2259 (4)	0.046 (1)
C4	0.1711 (2)	0.3130 (6)	0.2608 (4)	0.056 (1)
C5	0.2211 (2)	0.2993 (7)	0.1986 (5)	0.065 (2)
C6	0.2235 (2)	0.3826 (7)	0.0995 (5)	0.071 (2)
C7	0.1767 (2)	0.4750 (6)	0.0645 (5)	0.061 (1)
C8	−0.0137 (2)	0.7336 (5)	0.0588 (4)	0.044 (1)
C9	−0.0602 (2)	0.7914 (6)	0.1193 (4)	0.055 (1)
C10	−0.1021 (2)	0.8928 (6)	0.0698 (5)	0.060 (1)
C11	−0.0979 (2)	0.9379 (5)	−0.0408 (4)	0.053 (1)
C12	−0.0518 (2)	0.8844 (5)	−0.1008 (4)	0.053 (1)
C13	−0.0101 (2)	0.7835 (5)	−0.0513 (4)	0.050 (1)

Table 2. Selected geometric parameters (Å, °)

Co—O1	1.891 (3)	C4—C5	1.383 (7)
Co—N1	2.006 (4)	C5—C6	1.396 (8)
Br1—C11	1.892 (5)	C6—C7	1.348 (8)
N1—C1	1.287 (6)	C8—C13	1.387 (6)
N1—C8	1.428 (6)	C8—C9	1.401 (7)